

Atomic physics

Objects of this course:

This course aims to give students a broad perspective of the current understandings of :

- The structure of atoms, to the nucleus
- The interaction with one another
- The interaction with electric and magnetic fields.

Key Points:

- Building physics pictures
- Clear description of physics concepts

Teaching outlines

The following sections are included:

❖ Introduction (2 hour)

Motivations , why and how?

❖ Chapter 1.Theory of Relativity (6 hours)

1.1 Special relativity

1.2 General relativity

❖ Chapter 2. Basic aspects of Atom (Rutherford) (6 hours)

- 2.1 The background
- 2.2 Emergence of the Rutherford model
- 2.3 Rutherford Scattering formula
- 2.4 Experimental verification of the Rutherford Formula
- 2.5 Summary of the Significances and difficulties of the Nuclear model

❖ Chapter 3. Bohr model of the Hydrogen atom (6 hours)

- 3.1 Background
- 3.2 Bohr model
- 3.3 Experimental evidence I: Spectra
- 3.4 Experimental evidence II: Frank-Herz Experiment
- 3.5 Extension of the Bohr Model
- 3.6 Summary

❖ Chapter 4. Fine structure (Spin) (6 hours)

- 4.1. Magnetic moment produced by the electron orbital motion in an atom
 - 4.2. The Stern-Gerlach experiment
 - 4.3. The hypothesis of electron spin
 - 4.4. Doublet lines of Alkali Metals
 - 4.5. The Zeeman Effects (I)
 - 4.5. The Zeeman Effects (II)
 - 4.6. Summary of the Hydrogen energy spectrum
- Summary, Exercise and problems

❖ Chapter 5. Many-electron atoms (Pauli-Blocking) (6 hours)

- 5.1. The spectra and energy levels of Helium
- 5.2. The coupling of the two electrons
- 5.4. The periodic Table of elements
- 5.5. Summary, exercise and problems

❖ Chapter 6. X-Ray (6 hours)

- 6.1. The discovery of X-rays and their wave nature
- 6.2. Mechanisms for producing X-rays
- 6.3. Compton Scattering

6.4. Absorption of X-rays

6.5 Summary, problems

The curriculum form

- Lectures (by prof. Defu Fou)
- Seminars (students presentations)
- Summary & Exercise (prof. &TA)

Includes lectures and exams, and involves presentation and discussion

Evaluation of the course

Attendance & participation (10%)

Homework (20%)

Mid-term (25%)

Final test (45%)

Textbooks & References

(1) Modern atomic and Nuclear physics , by F. J Yang J.H Hamilton , 2010
World scientific Publishing Co. **(textbook)**

(2) The Physics of Atoms and Quanta, sixth edition, by H. Haken and H.
C. Wolf, 2003, published by Springer-Verlag and World scientific publ.

(3) Theoretical Atomic Physics, by H Friedrich , 2006 by Springer

(4) Atomic physics, by FP Miller, AF Vandome, J Mcbrewster, Alphascript
Publishing, 2009

(5) Riepe D. Atomic Physics and Human Knowledge, by Niels Bohr[M]// Atomic
Physics and Human Knowledge. И з д - в о И н о с т р а н н о й
л и т е р а т у р ы, 1961:596-597.

(6) Using Diode-Lasers for Atomic Physics, CE Wieman, L
Hollberg - 《Review of Scientific Instruments》, 1991

- (7) Atomic physics with super-high intensity lasers , M Protopapas, CH Keitel, PL Knight - 《Reports on Progress in Physics》 - 1997
- (8) Atomic Physics and Human Knowledge by Niels Bohr, D Riepe - Atomic Physics and Human Knowledge. - 1961
- (9) Discussion with Einstein on epistemological problems in atomic physics N Bohr - 《Schilpp P. a. albertnstein Philosopherscientist》 - 1949
- (10) A learning pathway in high - school level quantum atomic physics Juergen Petri, Hans Niedderer - 《International Journal of Science Education》>> 2012